

# Enantioselective Synthesis of (7*R*,8*R*,8*aR*)- and (7*S*,8*S*,8*aS*)-7-Hydroxy-8-indolizidinemethanol by 1,3-Dipolar Cycloaddition of 1-Pyrroline *N*-Oxide to Chiral Pentenoates

Franca Maria Cordero,<sup>\*,[a]</sup> Cristina Faggi,<sup>[a]</sup> Francesco De Sarlo,<sup>[a]</sup> and Alberto Brandi<sup>\*,[a]</sup>

**Keywords:** Chiral auxiliaries / Cycloadditions / Diastereoselectivity / Nitrogen heterocycles / Nitrones

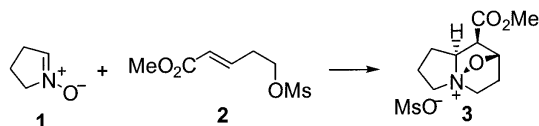
The 1,3-dipolar cycloadditions of pyrroline *N*-oxide (**1**) with (1*R*,2*S*)-*trans*-2-phenylcyclohexyl- and with (1*R*,2*S*,5*R*)-8-phenylmenthyl-pent-2-enoates (**9** and **10**) proceed with op-

posite diastereoselectivities. The two enantiomers of a new dihydroxyindolizidine, (+)-**6** and (–)-**6**, were synthesized using **4** and **5**, respectively, as the chiral auxiliary.

## Introduction

The 1,3-dipolar cycloaddition reaction of nitrones to alkenes has been successfully applied to the synthesis of a wide variety of natural and unnatural target molecules.<sup>[1]</sup> One of the main outstanding features of this type of concerted process is its ability to generate up to three new, adjacent stereogenic centres in a stereospecific manner. Moreover, the isoxazolidine cycloadducts can be further elaborated to provide polyfunctionalized cyclic or acyclic chiral compounds with complete control of the relative stereochemistry.

Nitrones add to (*E*)- $\beta$ -monosubstituted  $\alpha,\beta$ -unsaturated esters (e.g. methylcrotonate) in a regioselective manner, affording almost exclusively isoxazolidines substituted with the ester group at C-4. Despite the abundant literature available on the subject<sup>[2,3]</sup> and the numerous studies of asymmetric induction in cycloadditions to optically pure acrylates,<sup>[4]</sup> the behaviour of nitrones towards chiral (*E*)- $\beta$ -monosubstituted  $\alpha,\beta$ -unsaturated esters or amides has received little attention.<sup>[4c–4d,5–7]</sup>



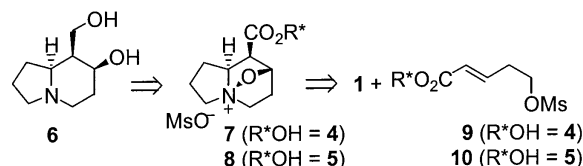
Scheme 1

Recently we have comprehensively revisited the 1,3-dipolar cycloaddition of the cyclic nitrone **1** with methyl (*E*)-5-[(methanesulfonyl)oxy]-2-pentenoate **2** (Scheme 1).<sup>[3]</sup> In connection with our project dealing with the synthesis of new enantiopure hydroxyindolizidines,<sup>[8]</sup> we carried out a study of the control of absolute stereochemistry in cycloadditions to chiral pentenoates for the synthesis of optically pure cycloadducts such as **3**. As a source of asymmetric induction, we chose two commercially available chiral aux-

iliaries: (–)-(1*R*,2*S*)-*trans*-2-phenylcyclohexanol<sup>[9]</sup> (**4**) and (–)-(1*R*,2*S*,5*R*)-8-phenylmenthol<sup>[10]</sup> (**5**), in view of their successful employment in asymmetric synthesis.<sup>[11]</sup>



In this paper we present the results of our study into the 1,3-dipolar cycloaddition reaction of 3,4-dihydro-2*H*-pyrrole 1-oxide (**1**) with the chiral pentenoates **9** and **10**, along with the transformation of the cycloadducts **7** and **8** into the new optically pure dihydroxyindolizidines (+)-**6** and (–)-**6** (Scheme 2).



Scheme 2

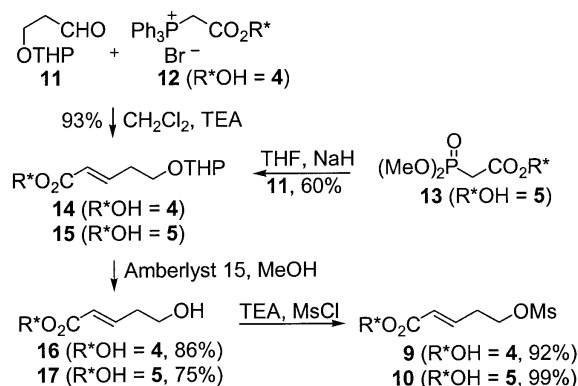
## Results and Discussion

The required chiral dipolarophiles **9** and **10** were prepared in good yields by condensation of the aldehyde **11**<sup>[12]</sup> either with the ylide generated in situ from the triphenylphosphonium bromide **12**, or with dimethylphosphonoacetate **13**, respectively (Scheme 3). The THP (tetrahydropyranyl) protecting group was replaced by the Ms (methanesulfonyl) group by treatment with Amberlyst<sup>®</sup>15 in methanol, followed by mesylation (Scheme 3).

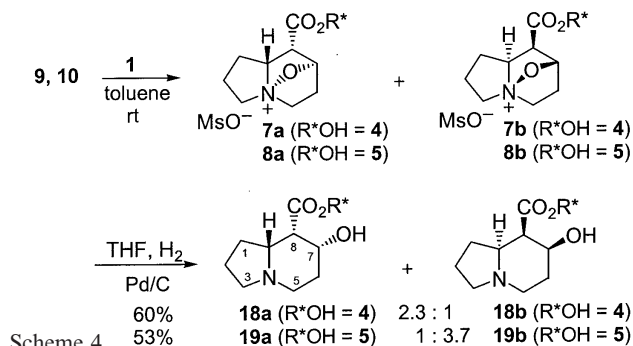
Reaction of **9** with **1** in toluene at room temperature gave the two diastereomeric salts **7a** and **7b** (Scheme 4). Hydrogenation of this mixture with Pd/C catalyst afforded the indolizidines **18a** and **18b** in 2.3:1 ratio (39% *de*), with an overall yield of 60% relative to **9** (Scheme 4).

The cycloaddition of **10** and **1** gave a white precipitate (30% yield) that was found to be the salt **8b** in analytically pure form. This was quantitatively transformed into indolizidinol **19b** by hydrogenation (Scheme 4). Similar treatment

<sup>[a]</sup> Dipartimento di Chimica organica “U. Schiff”, and Centro dei Composti Eterociclici-CNR, Università di Firenze, via G. Capponi 9, 50121 Firenze, Italy  
Fax: (internat.) + 39-055/2476964  
E-mail: cordero@chimorg.unifi.it



Scheme 3



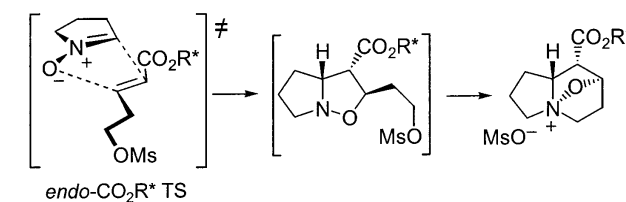
Scheme 4

of the mother liquor gave, after purification, the two indolizidines **19a** and **19b** in an overall molar ratio of 1:3.7 (57% *de*) and 53% yield relative to **10** (Scheme 4).

Only the salt **8b** could be completely analyzed by spectroscopic means. Its  $^1\text{H}$  NMR spectrum was in agreement with the presence of a positive charge on the nitrogen, inducing a deshielding effect on the 8a-H and 7-H protons ( $\delta_{8a\text{-H}} = 4.92$  and  $\delta_{7\text{-H}} = 5.17$ ), in analogy with the data previously found for the methyl ester **3** ( $\text{R} = \text{Me}$ ).<sup>[3]</sup> The relative configurations of the indolizidine moieties of **7a** and **b**, and **8a** and **b** were derived directly from those of the corresponding hydrogenated compounds **18a** and **b**, and **19a** and **b**, respectively. The indolizidine structures were assigned on the basis of the spectroscopic data and later confirmed by an X-ray analysis. The significant correlation between the IR and NMR spectra of **18a** and **b**, and **19a** and **b** with those of the hydrogenation product of **3**<sup>[3]</sup> suggested a correspondence of the relative stereochemistry of the three stereogenic centres on the indolizidine system. Particularly, the shape of the signal due to the resonance of 7-H in the  $^1\text{H}$  NMR spectra of **18a**, **19a**, and **19b** ( $\delta = 3.61$ , dt,  $J = 10.2$ ; 5.1 Hz,  $\delta = 3.49$ , br s, half-height width 21 Hz and  $\delta = 3.61$ , br s, half-height width 22 Hz, respectively) was consistent with the assigned stereochemistry. Moreover, the presence of Bohlmann bands<sup>[13]</sup> in the IR spectrum of **18a**, **18b**, and **19b** (2795, 2736  $\text{cm}^{-1}$ , 2795, 2740  $\text{cm}^{-1}$  and 2790, 2736  $\text{cm}^{-1}$ , respectively) showed that the ring systems were *trans*-fused.

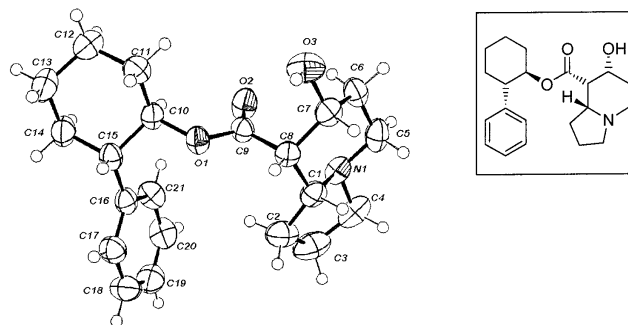
The observed stereochemistry of the indolizidine moieties attested that the diastereoisomers **7a** and **b**, and **8a** and **b** had originated from an *endo*- $\text{CO}_2\text{R}^*$  approach of the two

reagents **1** and either **9** or **10** in the cycloaddition reaction (Scheme 5).



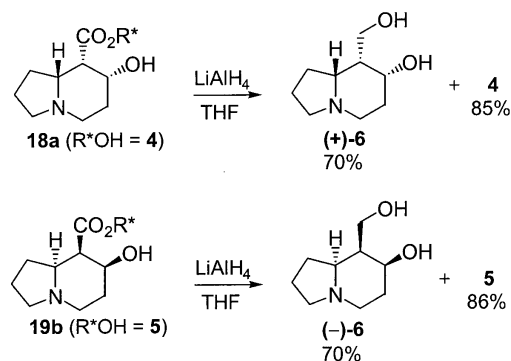
Scheme 5

The absolute stereochemistry of the indolizidine moiety in the major product **18a**, deriving from (1*R*,2*S*)-2-phenylcyclohexanol (**4**), was assigned as (7*R*,8*S*,8a*R*) by X-ray analysis of a crystal obtained by slow evaporation of a  $\text{CH}_3\text{CN}$  solution of **18a** (Figure 1).

Figure 1. ORTEP drawing from the X-ray crystal structure of diastereoisomer **18a**

The assignment of absolute stereochemistry to the compounds **19a** and **19b** deriving from **5** could be assessed indirectly by further transformation of the esters **18** and **19**.

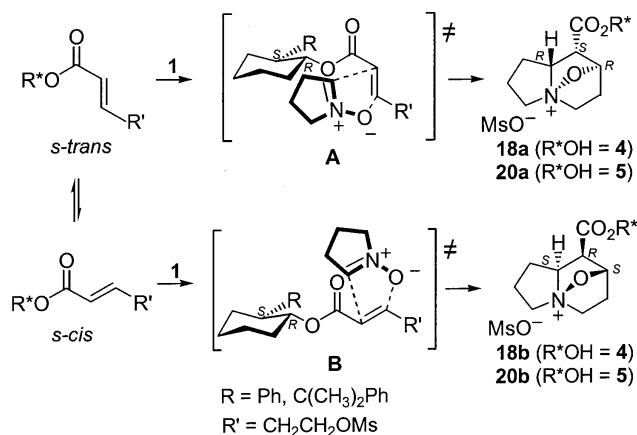
The two main products **18a** and **19b** were reduced with  $\text{LiAlH}_4$  to dihydroxyindolizidine **6**. Unexpectedly, the two enantiomers (+)-**6** ( $[\alpha]_{\text{D}}^{24} = +30.4$ ) and (−)-**6** ( $[\alpha]_{\text{D}}^{24} = -30.1$ ) were obtained from **18a** and **19b**, respectively (Scheme 7), showing that the absolute configurations of the indolizidine moieties in the main cycloadducts **7a** and **8b** were opposite (Scheme 6). In both cases, the chiral auxiliaries **4** and **5** were recovered in good yield (Scheme 6).



Scheme 6

The observed stereochemical outcome of **18a** was consistent with a preferential attack of the nitrone **1** at the less sterically hindered  $\pi$ -face (2*Si*,3*Re* face) of the *s-trans* conformer<sup>[14]</sup> of **9** (**A**, Scheme 7). The moderate stereocontrol

(39% *de*) induced by *trans*-2-phenylcyclohexanol **4** can be ascribed to the “widening V” relative arrangement<sup>[15]</sup> of the phenyl and the enoate moieties, and to the occurrence of addition also on the *s-cis* conformer, having the opposite face (*2Re,3Si*) as the more accessible one (Scheme 7).



Scheme 7

The higher degree of diastereoselectivity (57% *de*) shown by 8-phenylmenthol **5**, (compared to **4**) is in agreement with other examples of uncatalyzed cycloaddition reactions of 8-phenylmenthol derivatives.<sup>[16]</sup> The efficiency of phenylmenthol has been ascribed to the prevalence of the stacked conformation in the corresponding crotonate.<sup>[11]</sup> Actually, the olefinic proton resonances in the pentenoate **10** proved to be very close to those in 8-phenylmenthol crotonate<sup>[11b]</sup> ( $\delta = 6.33$  and  $5.32$  and  $\delta = 6.44$  and  $5.32$ , respectively) and shifted upfield compared to those in **9** ( $\delta = 6.65$  and  $5.70$ ), suggesting the presence of **10** in the stacked geometry. The opposite senses of diastereoselectivity showed by the two chiral auxiliaries **4** and **5** cannot easily be rationalised. In general, the two chiral auxiliaries afford the same absolute stereochemistry of products in the reaction studied.<sup>[11]</sup> However, at least one other example of this dichotomy has been reported, in a nucleophilic addition reaction to glyoxylate esters.<sup>[11a]</sup>

Assuming a high degree of  $\pi$ -face shielding in the cycloaddition of **1** and **10**, the experimental results proved that the preferred conformation of the pentenoate **10** in the transition state is *s-cis* (**B**, Scheme 7) and suggested that the moderate diastereoselectivity was determined by a low-level rotamer control.

## Conclusion

We have proved that the 1,3-dipolar cycloadditions between the nitron **1** and the pent-2-enoates **9** and **10** occur with a moderate degree of diastereoselectivity; higher when the chiral auxiliary is (–)-(1*R,2S,5R*)-8-phenylmenthol (**5**) than when it is (–)-(1*R,2S*)-*trans*-2-phenylcyclohexanol (**4**). The observation that the two chiral auxiliaries act in a complementary fashion is remarkable. The diastereoselectivity observed in the cycloaddition is much lower than the induction showed by the same auxiliaries in other types of reac-

tion,<sup>[11]</sup> but is interesting in the context of the few data reported for chiral crotonates in nitron cycloadditions. In fact, Murahashi et al.<sup>[5b]</sup> reported a lower diastereoselectivity in the 1,3-dipolar cycloaddition between cyclic nitrones and chiral crotonates in the absence of Lewis acid, while the introduction of greater than stoichiometric amounts of  $\text{ZnI}_2$  increased the diastereoselectivity, but reduced the reaction yields. Finally, very fair diastereoselectivity and yield were recently obtained in metal-catalyzed cycloadditions of crotonyloxazolidinones,<sup>[5d,5e,17]</sup> but only acyclic and *C,N*-aromatic nitrones were investigated and therefore the data are less readily comparable.

The two diastereoisomer pairs **18** and **19** obtained by reduction of the primary cycloadducts **7** and **8**, respectively, are easily separated by column chromatography; therefore our strategy represents a convenient access to both enantiomeric new dihydroxyindolizidines (+)-**6** and (–)-**6** in optically pure form.

## Experimental Section

**General Remarks:** All reactions that required dry conditions were performed under a nitrogen atmosphere using anhydrous solvents. – Melting points (mp) are uncorrected. –  $R_f$  values refer to TLC on 0.25 mm silica gel plates (Merck F254) using the same eluent used for the chromatographic separation of the compound unless otherwise stated. –  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra (in  $\text{CDCl}_3$ , unless otherwise stated) were recorded on a Varian Gemini ( $^1\text{H}$  200 MHz) or Bruker DRX 500 ( $^1\text{H}$  500 MHz). Chemical shifts are given in ppm. Notation s, d, t, q, m, and br indicate singlet, doublet, triplet, quadruplet, multiplet, and broad, respectively. – IR spectra (in  $\text{CDCl}_3$ , unless otherwise stated) were recorded with a Perkin–Elmer 881 spectrophotometer. – Mass spectra (MS) were recorded on 5792A Hewlett-Packard and QMD 1000 Carlo Erba instruments. – Microanalyses were measured with a Perkin–Elmer 240 C instrument.

**(1*R,2S*)-*trans*-2-Phenyl-1-cyclohexyl (E)-5-[(Tetrahydro-2*H*-pyran-2-yl)oxy]-2-pentenoate (14):** Triethylamine (TEA, 6.8 mmol, 691 mg) was added to a solution of **12** ( $[\alpha]_D^{20} = +6.8$  ( $c = 1.25$ ,  $\text{CHCl}_3$ ), 5.7 mmol, 3.20 g) in  $\text{CH}_2\text{Cl}_2$  (10 mL). The solution was stirred at room temp. for 2.5 h and then a solution of **11** (11.4 mmol, 1.80 g) in  $\text{CH}_2\text{Cl}_2$  (5 mL) was added. The mixture was heated at reflux temperature for 3 h, kept at room temp. for 60 h, filtered through a short pad of Celite and concentrated to obtain a yellow oil (2.5 g). The crude oil was purified by column chromatography on silica gel (eluent  $\text{CHCl}_3/\text{MeOH}$ , 50:1) to obtain **14** (1.90 g, 93%) as an inseparable mixture of the two diastereoisomers.

**Compound 14:**  $R_f = 0.64$ . –  $^1\text{H}$  NMR (diastereomeric mixture):  $\delta = 7.30$ – $7.16$  (m, 5 H), 6.75 (dt,  $J = 15.5$ ; 7.6 Hz, 1 H), 5.65 (d, 15.5 Hz, 1 H), 5.15–4.95 (m, 1 H), 4.55 (br s, 1 H), 3.82–3.65 (m, 2 H), 3.50–3.38 (m, 2 H), 2.80–2.60 (m, 1 H), 2.42–2.30 (m, 2 H), 2.20–1.25 (m, 14 H). –  $^{13}\text{C}$  NMR (major diastereoisomer):  $\delta = 165.6$  s, 145.1 d, 143.1 s, 128.2 d (2 C), 127.4 d (2 C), 126.2 d, 122.9 d, 98.6 d, 75.7 d, 65.4 t, 62.9 t, 49.6 d, 34.0 t, 32.4 t, 32.3 t, 30.5 t, 25.8 t, 25.4 t, 24.7 t, 19.4 t. – MS  $m/z$  (rel intensity): 244 (4), 158 (100), 129 (15), 99 (18), 91 (41), 85 (88). – IR:  $\tilde{\nu}$  [ $\text{cm}^{-1}$ ] = 3031, 2939, 1708, 1656, 1601, 1280, 1176.

**(1*R,2S,5R*)-8-Phenylmenthyl (E)-5-[(Tetrahydro-2*H*-pyran-2-yl)oxy]-2-pentenoate (15):** A solution of **13** ( $[\alpha]_D^{23} = +9.6$  ( $c = 1.1$ ,

MeOH), 3 mmol, 1.160 g) in THF (10 mL) was added dropwise to NaH (3.45 mmol, 78 mg) at 0 °C. The mixture was stirred at room temp. for 1 h, then cooled at 0 °C, and a solution of **11** (545 mg, 3.45 mmol) in THF (3 mL) was added dropwise. The mixture was kept at room temp. for 15 h, warmed at 40 °C for 3 h and then diluted with diethyl ether and washed with H<sub>2</sub>O. The aqueous phase was extracted 3 times with diethyl ether and the combined extracts were washed with sat. aq. NaHCO<sub>3</sub>, dried with Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The crude yellow oil (1.5 g) was purified by column chromatography (eluent: petroleum ether/AcOEt 6:1) to obtain **15** (726 mg, 60%) as an inseparable mixture of the two diastereoisomers, in addition to some unchanged **13** (200 mg).

**Compound 15:**  $R_f = 0.47$ . – <sup>1</sup>H NMR (diastereomeric mixture):  $\delta = 7.30$ – $7.20$  (m, 4 H),  $7.18$ – $7.05$  (m, 1 H),  $6.53$  (m, 1 H),  $5.23$  (m, 2 H),  $4.82$  (m, 1 H),  $4.58$  (m, 1 H),  $3.86$ – $3.72$  (m, 2 H),  $3.58$ – $3.38$  (m, 2 H),  $2.38$  (m, 2 H),  $2.10$ – $1.65$  (m, 4 H),  $1.62$ – $1.38$  (m, 6 H),  $1.36$ – $1.18$  (m, 1 H),  $1.31$  (s, 3 H),  $1.21$  (s, 3 H),  $1.15$ – $0.80$  (m, 5 H).

**(1R,2S)-trans-2-Phenyl-1-cyclohexyl (E)-5-Hydroxy-2-pentenoate (16):** A solution of the diastereomeric mixture of **14** (5.3 mmol, 1.90 g) in MeOH (10 mL) was warmed to 40 °C for 6.5 h in the presence of Amberlyst® 15 (160 mg). The mixture was filtered through a short pad of Celite, concentrated under reduced pressure, and the crude product was purified by column chromatography (eluent: CHCl<sub>3</sub>/MeOH 50:1) to obtain **16** (1.27 g, 86%) in analytically pure form.

**Compound 16:**  $R_f = 0.2$ . – <sup>1</sup>H NMR:  $\delta = 7.31$ – $7.11$  (m, 5 H),  $6.70$  (dt,  $J = 15.7$ ;  $7.1$  Hz, 1 H),  $5.68$  (dt,  $J = 15.6$ ;  $1.5$ , 1 H),  $5.03$  (dt,  $J = 4.8$ ;  $10.6$  Hz, 1 H),  $3.72$ – $3.60$  (m, 2 H),  $2.71$  (dt,  $J = 3.7$ ;  $11.7$  Hz, 1 H),  $2.35$  (dq,  $J = 1.5$ ;  $6.5$  Hz, 2 H),  $2.23$ – $2.10$  (m, 1 H),  $2.02$ – $1.26$  (m, 8 H). – <sup>13</sup>C NMR:  $\delta = 165.6$  s,  $144.6$  d,  $143.2$  s,  $128.2$  d (2 C),  $127.5$  d (2 C),  $126.3$  d,  $123.7$  d,  $75.9$  d,  $60.9$  t,  $49.7$  d,  $35.3$  t,  $34.0$  t,  $32.3$  t,  $25.8$  t,  $24.7$  t. MS  $m/z$  (rel intensity): 177 (3), 158 (88), 143 (10), 130 (14), 117 (7), 99 (27), 91 (56), 81 (100). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 3626, 3033, 2939, 1707, 1654, 1448, 1272, 1190, 1021. –  $[\alpha]_D^{25} = -55.6$  ( $c = 1.36$ , MeOH). – C<sub>17</sub>H<sub>22</sub>O<sub>3</sub> (274.4): calcd. C 74.42, H 8.08; found C 74.61, H 8.09.

**(1R,2S,5R)-8-Phenylmenthyl (E)-5-Hydroxy-2-pentenoate (17):** Following the procedure described above, the diastereomeric mixture of **15** (1.65 mmol, 685 mg) gave **17**<sup>[18]</sup> (400 mg, 75%).

**Compound 17:** <sup>1</sup>H NMR:  $\delta = 7.29$ – $7.20$  (m, 4 H),  $7.15$ – $7.06$  (m, 1 H),  $6.44$  (dt,  $J = 15.6$ ;  $7.0$  Hz, 1 H),  $5.37$ – $5.28$  (m, 1 H),  $4.48$  (td,  $J = 10.8$ ;  $4.4$  Hz, 1 H),  $3.69$  (t,  $J = 6.4$  Hz, 2 H),  $2.38$ – $2.26$  (m, 2 H),  $2.12$ – $1.99$  (m, 1 H),  $1.96$ – $1.84$  (m, 1 H),  $1.78$ – $1.59$  (m, 3 H),  $1.52$ – $1.40$  (m, 1 H),  $1.30$  (s, 3 H),  $1.29$ – $1.20$  (m, 1 H),  $1.20$  (s, 3 H),  $1.18$ – $0.92$  (m, 2 H),  $0.86$  (d,  $J = 6.6$  Hz, 3 H). – <sup>13</sup>C NMR:  $\delta = 165.3$  s,  $151.8$  s,  $144.1$  d,  $127.9$  d (2 C),  $125.4$  d (2 C),  $124.7$  d,  $124.0$  d,  $74.2$  d,  $60.8$  t,  $50.4$  d,  $41.7$  t,  $39.6$  s,  $35.3$  t,  $34.6$  t,  $31.3$  d,  $28.2$  q,  $26.5$  t,  $24.6$  q,  $21.8$  q. – MS:  $m/z$  (rel. intensity): 330 [M<sup>+</sup>] (1), 214 (21), 211 (11), 199 (5), 119 (98), 118 (100), 99 (64), 91 (88). –  $[\alpha]_D^{25} = -6.4$  ( $c = 0.96$ , MeOH). – C<sub>21</sub>H<sub>30</sub>O<sub>3</sub> (330.5): calcd. C 76.33, H 9.15; found C 76.44, H 9.50.

**(1R,2S)-trans-2-Phenyl-1-cyclohexyl (E)-5-[(Methylsulfonyl)oxy]-2-pentenoate (9):** MsCl (82 mg, 0.71 mmol) was added dropwise to a solution of **16** (150 mg, 0.55 mmol) and TEA (71 mg 0.71 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3 mL), with cooling in an ice/water bath. The mixture was allowed to stand at room temp. for 1.5 h and then treated with ice. The separated organic phase was washed sequentially with 1 M HCl, sat. aq. NaHCO<sub>3</sub> and H<sub>2</sub>O, and dried with Na<sub>2</sub>SO<sub>4</sub>. The sol-

vent was removed under reduced pressure to give **9** (184 mg, 96%), which was used in the next step without further purification.

**Compound 9:** <sup>1</sup>H NMR:  $\delta = 7.79$ – $7.13$  (m, 5 H),  $6.65$  (dt,  $J = 15.7$ ;  $7.0$  Hz, 1 H),  $5.70$  (dt,  $J = 15.7$ ;  $1.5$  Hz, 1 H),  $5.04$  (dt,  $J = 4.7$ ;  $10.3$  Hz, 1 H),  $4.24$  (t,  $J = 6.4$  Hz, 2 H),  $2.93$  (s, 3 H),  $2.71$  (dt,  $J = 3.4$ ;  $11.2$  Hz, 1 H),  $2.54$  (dq,  $J = 1.5$ ;  $6.6$  Hz, 2 H),  $2.21$ – $2.16$  (m, 1 H),  $2.05$ – $1.22$  (m, 7 H). – <sup>13</sup>C NMR:  $\delta = 165.0$  s,  $143.1$  s,  $141.5$  d,  $128.2$  d (2 C),  $127.4$  d (2 C),  $126.3$  d,  $124.7$  d,  $76.1$  d,  $67.2$  t,  $49.6$  d,  $37.5$  q,  $34.0$  t,  $32.3$  t,  $31.6$  t,  $25.8$  t,  $24.7$  t. – MS  $m/z$  (rel intensity): 176 (67), 158 (11), 143 (15), 130 (56), 117 (30), 104 (40), 91 (100), 77 (20). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 2940, 2865, 1711, 1449, 1358, 1174.

**(1R,2S,5R)-8-Phenylmenthyl (E)-5-[(Methylsulfonyl)oxy]-2-pentenoate (10):** Following the procedure described above, the alcohol **17** (340 mg, 1.0 mmol) gave **10** (415 mg, 99%) as an oil, which was used in the next step without further purification.

**Compound 10:** <sup>1</sup>H NMR:  $\delta = 7.26$ – $7.10$  (m, 4 H),  $7.16$ – $7.11$  (m, 1 H),  $6.33$  (dt,  $J = 15.9$ ;  $6.8$  Hz, 1 H),  $5.32$  (dt,  $J = 15.9$ ,  $1.4$  Hz, 1 H),  $4.84$  (dt,  $J = 4.2$ ;  $10.7$  Hz, 1 H),  $4.24$  (t,  $J = 6.6$  Hz, 2 H),  $3.01$  (s, 3 H),  $2.50$  (qd,  $J = 1.4$ ;  $6.6$  Hz, 2 H),  $2.14$ – $2.01$  (m, 1 H),  $1.96$ – $0.90$  (m, 7 H),  $1.29$  (s, 3 H),  $1.20$  (s, 3 H),  $0.87$  (d,  $J = 6.2$  Hz, 3 H). – <sup>13</sup>C NMR:  $\delta = 164.9$  s,  $151.8$  s,  $141.0$  d,  $128.0$  d (2 C),  $125.4$  d (2 C),  $124.8$  d,  $124.7$  d,  $74.4$  d,  $67.1$  t,  $50.4$  d,  $41.6$  t,  $39.6$  s,  $37.6$  q,  $34.6$  t,  $31.6$  t,  $31.2$  q,  $28.3$  q,  $26.5$  t,  $24.5$  d,  $21.8$  q. – MS  $m/z$  (rel intensity): 243 (2), 214 (6), 177 (4), 158 (23), 119 (100), 91 (37). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 2927, 1704, 1651, 1517, 1448, 1357, 1173.

**(7R,8S,8aR)-7-Hydroxy-8-[(1R,2S)-trans-2-phenyl-1-cyclohexyloxycarbonyl]octahydroindolizine (18a)<sup>[19]</sup> and (7S,8R,8aS)-7-Hydroxy-8-[(1R,2S)-trans-2-phenyl-1-cyclohexyloxycarbonyl]octahydroindolizine (18b):** A solution of **1** (391 mg, 4.6 mmol) in toluene (11 mL) was added dropwise to **9** (810 mg, 2.3 mmol), with cooling in an ice/water bath. The reaction mixture was stirred at 0 °C for 1 h and then at room temp. for 4 days. The solvent was removed under reduced pressure to give a dark viscous oil. The crude mixture was diluted with THF (10 mL) and hydrogenated over 10% Pd-C (49 mg, 46  $\mu$ mol) at atmospheric pressure for 3 days. The catalyst was removed by filtration through a short pad of Celite. The filtrate was concentrated, diluted with CH<sub>2</sub>Cl<sub>2</sub>, treated with Amberlyst® A26 for 5 min and filtered through a short pad of Celite. The concentrated crude mixture of the two diastereoisomers **18a** and **18b** in 2.3:1 ratio was purified by column chromatography on silica gel (eluent: initially CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 15:1, then the polarity was gradually increased) to afford analytically pure **18a** (213 mg) and **18b** (93 mg) with a total yield of 60% relative to the mesylate **9**.

**Compound 18a:**  $R_f = 0.20$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 5:1). m.p. 168 °C (benzene). – <sup>1</sup>H NMR:  $\delta = 7.30$ – $7.16$  (m, 5 H),  $5.07$  (dt,  $J = 4.2$ ;  $10.5$  Hz, 1 H),  $3.61$  (dt,  $J = 10.2$ ;  $5.1$  Hz, 1 H),  $2.99$  (dt,  $J = 11.3$ ,  $4.2$  Hz, 1 H),  $2.86$  (m, 2 H),  $2.70$  (dt,  $J = 3.5$ ;  $11.3$  Hz, 1 H),  $2.19$ – $0.70$  (m, 18 H). – <sup>13</sup>C NMR:  $\delta = 171.1$  s,  $143.4$  s,  $128.5$  d (2 C),  $127.5$  d (2 C),  $126.5$  d,  $76.7$  d,  $69.6$  d,  $63.5$  d,  $53.7$  t,  $49.7$  d,  $49.4$  t,  $49.0$  d,  $34.6$  t,  $32.4$  t,  $30.7$  t,  $25.8$  t (2 C),  $24.6$  t,  $21.5$  t. – MS:  $m/z$  (rel. intensity): 343 [M<sup>+</sup>] (7), 184 (100), 166 (12), 140 (28), 122 (41), 96 (54), 91 (42). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 3562, 2939, 2860, 2795, and 2736 (Bohlmann bands),<sup>[13]</sup> 1710, 1447, 1326, 1258, 1180. – C<sub>21</sub>H<sub>29</sub>NO<sub>3</sub> (343.5): calcd. C 73.44, H 8.51, N 4.08; found C 73.00, H 8.30, N 4.30. –  $[\alpha]_D^{27} = +29.0$  ( $c = 0.5$ , MeOH).

**X-ray Structural Analysis:** Orthorhombic, space group  $P2_12_12_1$ ,  $a = 5.398(5)$ ,  $b = 12.223(5)$ ,  $c = 28.801(5)$  Å,  $V = 1900.0(2)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.200$ ,  $\mu = 0.630$  mm<sup>−1</sup>,  $F(000) = 744$ . Analysis on a



prismatic transparent single crystal was carried out with a Siemens P4 X-ray diffractometer at room temperature. Graphite-monochromated Cu-K $\alpha$  radiation was used for cell parameter determination and data collection. The intensities of two standard reflections were monitored during data collection to check the stability of the crystal: no loss of intensity was recognized. The integrated intensities, measured using the  $\theta/2\theta$  scan mode, were corrected for Lorentz and polarization effects.<sup>[20]</sup> The number of reflections collected was 2017 with a  $3.07 < \theta < 55.00$  range; 1829 were independent and the final *R* index was 0.0404 for reflections having  $I > 2\sigma I$ , and 0.0423 for all data. The non-hydrogen atoms were refined anisotropically; aromatic and methylene hydrogens were assigned in calculated positions, the others were found by Fourier difference synthesis; all of them were refined as isotropic. This structure was solved by SIR92<sup>[21]</sup> direct methods and refined using the full-matrix, least-squares method on  $F^2$  provided by SHELXL97.<sup>[22]</sup>

**Compound 18b:**  $R_f = 0.32$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 5:1). m.p. 52–53 °C. – <sup>1</sup>H NMR (500 MHz):  $\delta = 7.30$ –7.18 (m, 5 H), 5.12 (dt,  $J = 4.4$ ; 10.6 Hz, 1 H), 3.49 (br s, 1 H), 3.03–2.98 (m, 2 H), 2.78 (br s, 1 H), 2.69 (dt,  $J = 3.3$ ; 11.6 Hz, 1 H), 2.22–0.90 (m, 18 H). – <sup>13</sup>C NMR:  $\delta = 170.6$  s, 142.9 s, 128.4 d (2 C), 127.6 d (2 C), 126.6 d, 76.0 d, 10.0 d, 63.9 d, 53.9 t, 50.0 d, 49.6 t, 48.4 d, 34.0 t, 32.6 t, 30.6 t, 26.6 t, 25.7 t, 24.7 t, 21.7 t. – MS;  $m/z$  (rel. intensity): 343 [ $M^+$ ] (3), 184 (100), 166 (9), 140 (13), 122 (28), 96 (25), 91 (19). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 3562, 2930, 2857, 2795 and 2740 (Bohlmann bands), 1708, 1359, 1221. – C<sub>21</sub>H<sub>29</sub>NO<sub>3</sub> (343.5): calcd. C 73.44, H 8.51, N 4.08; found C 73.09, H 8.75, N 4.11. –  $[\alpha]_D^{26} = -39.1$  ( $c = 0.3$ , MeOH).

**(7*R*,8*R*,8*aR*)-7-Hydroxyindolizidin-8-ylmethanol [(+)-6]:** A suspension of LiAlH<sub>4</sub> (7.6 mg, 0.20 mmol) in THF (3 mL) was cooled in an ice/water bath and treated dropwise with a solution of **18a** (34 mg, 0.10 mmol) in THF (3 mL). The mixture was maintained at 40 °C for 4 h, then at room temp. for 14 h and finally cooled to 0 °C and treated sequentially with H<sub>2</sub>O (25  $\mu$ L), 10% aq. NaOH (50  $\mu$ L), and H<sub>2</sub>O (70  $\mu$ L). The heterogeneous system was vigorously stirred at room temp. overnight, filtered through a short pad of Celite, concentrated and purified by column chromatography on silica gel (eluent: CH<sub>2</sub>Cl<sub>2</sub>/MeOH/conc. NH<sub>4</sub>OH, 75: 24.5: 5) to afford pure (+)-**6** (12 mg, 70%) and the chiral auxiliary **4** (15 mg, 85%).

**Compound (+)-6:**  $R_f = 0.12$ . – <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O):  $\delta = 3.92$  (dt,  $J = 10.5$ ; 4.8 Hz, 1 H), 3.84 (A part of an ABX system,  $J = 10.0$ ; 5.0 Hz, 1 H), 3.76 (B part of an ABX system;  $J = 10.0$ ; 5.0 Hz, 1 H), 3.04 (dt,  $J = 11.7$ ; 4.3 Hz, 1 H), 2.99–2.95 (m, 1 H), 2.49 (b, 1 H), 2.31–2.16 (m, 2 H), 2.13 (m, 1 H), 1.88–1.73 (m, 5 H), 1.67–1.56 (m, 1 H). – <sup>13</sup>C NMR (CD<sub>3</sub>OD):  $\delta = 73.3$  d, 67.0 d, 59.0 t, 54.7 t, 51.3 t, 44.4 d, 31.8 t, 26.8 t, 22.8 t. – MS;  $m/z$  (rel. intensity): 171 [ $M^+$ ] (15), 170 (15), 154 (35), 140 (15), 122 (34), 112 (30), 96 (100), 84 (94), 83 (79). –  $[\alpha]_D^{24} = +30.4$  ( $c = 1.0$ , MeOH). – C<sub>9</sub>H<sub>17</sub>NO<sub>2</sub> (171.2): calcd. C 63.13, H 10.01, N 8.18; found C 62.86, H 10.04, N 8.33.

**(7*S*,8*S*,8*aR*)-4,7-Epoxy-8-[(1*R*,2*S*,5*R*)-8-phenylmenthyloxy-carbonyl]octahydroindolizinium Methanesulfonate (**8a**) and (7*R*,8*R*,8*aS*)-8-[(1*R*,2*S*,5*R*)-4,7-Epoxy-8-phenylmenthyloxy-carbonyl]octahydroindolizinium Methanesulfonate (**8b**):** A solution of **1** (170 mg, 2 mmol) in toluene (5 mL) was cooled in an ice/water bath and treated dropwise with a solution of **10** (408 mg, 1 mmol) in toluene (7 mL). The mixture was then stirred at room temp. for 15 days. The precipitate was filtered, washed with light petroleum ether and dried to give analytically pure **8b** (144 mg, 30%) as a white powder. A mixture of **8a** and **8b** (318 mg) of lower purity, was obtained from the mother liquor.

**Compound 8b:** <sup>1</sup>H NMR:  $\delta = 7.35$ –7.20 (m, 4 H), 7.10–7.05 (m, 1 H), 5.20–5.05 (m, 1 H), 5.17 (d,  $J = 4.0$  Hz, 1 H), 5.00–4.85 (m, 2 H), 4.75–4.58 (m, 1 H), 3.85–3.70 (m, 1 H), 3.63–3.45 (m, 1 H), 2.77 (m, 3 H), 2.70–1.95 (m, 5 H), 1.93–0.90 (m, 10 H), 1.28 (s, 3 H), 1.20 (s, 3 H), 0.86 (d,  $J = 6.3$  Hz, 3 H). – <sup>13</sup>C NMR:  $\delta = 166.7$  s, 152.0 s, 128.0 d (2 C), 125.6 d (2 C), 125.0 d, 83.7 d, 79.1 d, 75.1 d, 59.8 t, 57.7 t, 52.1 d, 49.9 d, 41.6 t, 39.4 q, 39.3 s, 34.1 t, 31.4 q, 31.3 q, 29.8 t, 26.9 t, 25.9 t, 24.6 t, 21.8 d, 20.9 q. – C<sub>26</sub>H<sub>30</sub>NO<sub>6</sub>S (493.7): calcd. C 63.26, H 7.96, N 2.84; found C 63.09, H 8.05, N 2.85.

**(7*R*,8*S*,8*aR*)-7-Hydroxy-8-[(1*R*,2*S*,5*R*)-8-phenylmenthyloxy-carbonyl]octahydroindolizine (**19a**) and (7*S*,8*R*,8*aS*)-7-Hydroxy-8-[(1*R*,2*S*,5*R*)-8-phenylmenthyloxy-carbonyl]octahydroindolizine (**19b**):** Salt **8b** (120 mg, 0.24 mmol) in THF (4 mL) was hydrogenated over 10% Pd-C (5.2 mg) at atmospheric pressure overnight. The catalyst was removed by filtration on a short pad of Celite, and the filtrate was concentrated, diluted with CH<sub>2</sub>Cl<sub>2</sub> and treated with Amberlyst® A26 for 15 min. The mixture was filtered through a short pad of Celite, dried with Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated to give analytically pure **19b** as a white solid (95 mg, 99%).

The crude mixture of **8a** and **8b** (318 mg) was hydrogenated separately to give a mixture of **19a** and **19b** as described above. The resulting crude solid was purified by column chromatography on silica gel (eluent: CH<sub>2</sub>Cl<sub>2</sub>/MeOH/conc. NH<sub>4</sub>OH, 150:10:1) to afford pure **19a** (45 mg) and **19b** (35 mg). Overall, the two diastereoisomers **19a** and **19b** were obtained in 3.7:1 ratio with a total yield of 53% relative to the mesylate **10**.

**Compound 19a:**  $R_f = 0.10$ . – <sup>1</sup>H NMR:  $\delta = 7.32$ –7.20 (m, 4 H), 7.18–7.10 (m, 1 H), 4.89 (dt,  $J = 4.0$ ; 11.0 Hz, 1 H), 3.49 (br s, half-height width 21 Hz, 1 H), 3.10–2.98 (m, 2 H), 2.52–0.83 (m, 19 H), 1.33 (s, 3 H), 1.21 (s, 3 H), 0.87 (d,  $J = 6.6$  Hz, 3 H).

**Compound 19b:**  $R_f = 0.33$ . m.p. 106–108 °C. – <sup>1</sup>H NMR (300 MHz):  $\delta = 7.35$ –7.21 (m, 4 H), 7.18–7.11 (m, 1 H), 4.83 (dt,  $J = 4.1$ ; 10.8 Hz, 1 H), 3.61 (br s, half-height width 22 Hz, 1 H), 3.09–2.96 (m, 2 H), 2.64 (br s, 1 H), 2.29–0.80 (m, 18 H), 1.32 (s, 3 H), 1.22 (s, 3 H), 0.88 (d,  $J = 6.5$  Hz, 3 H). – <sup>13</sup>C NMR (75 MHz):  $\delta = 171.8$  s, 151.6 s, 128.0 d (2 C), 125.5 d (2 C), 125.1 d, 74.9 d, 69.9 d, 64.0 d, 54.2 t, 50.4 d, 50.0 t, 48.1 d, 42.0 t, 39.6 s, 34.6 t, 31.4 q, 30.9 t, 29.7 t, 27.8 q, 26.6 t, 26.4 t, 25.3 d, 21.8 q. – MS;  $m/z$  (rel. intensity): 399 [ $M^+$ ] (4), 184 (100), 166 (9), 152 (6), 140 (6), 122 (15), 97 (24), 84 (36). – IR:  $\tilde{\nu}$  [cm<sup>−1</sup>] = 3568, 2960, 2790 and 2736 (Bohlmann bands),<sup>[13]</sup> 1699, 1597, 1492, 1455, 1441, 1371, 1324, 1172, 1084. –  $[\alpha]_D^{20} = -33.9$  ( $c = 0.92$ ; MeOH). – C<sub>25</sub>H<sub>37</sub>NO<sub>3</sub> (399.6): calcd. C 75.15, H 9.33, N 3.51; found C 75.09, H 9.43, N 3.15.

**(7*S*,8*S*,8*aS*)-7-Hydroxyindolizidin-8-ylmethanol ((-)-6):** Reduction of **19b** (80 mg, 0.20 mmol) following the procedure described above afforded pure (–)-**6** (24 mg, 70%) and the chiral auxiliary **5** (40 mg, 86%).

**Compound (–)-6:**  $[\alpha]_D^{24} = -30.1$  ( $c = 0.87$ , MeOH). The <sup>1</sup>H NMR, <sup>13</sup>C NMR, and MS spectra were identical with those recorded for (+)-**6**.

## Acknowledgments

The authors thank MURST (Ministry of University and Scientific and Technological Research, Rome, Italy) for financial support (Cofin 1998–2000). The assistance of Mr. Sandro Papaleo and Mrs. Brunella Innocenti for mass spectra and microanalyses is gratefully acknowledged.

- [1] [1a] J. J. Tufariello, *1,3-Dipolar Cycloaddition Chemistry* (Ed.: A. Padwa), John Wiley & Sons, Inc., New York, **1984**; vol. 2. Chapt. 9, pp. 83–168. — [1b] K. B. G. Torsell, *Nitrile Oxides, Nitrones, and Nitronates in Organic Synthesis*; VCH Verlagsgesellschaft: Weinheim, **1988**. — [1c] P. N. Confalone, E. M. Huie, *Organic Reactions* **1988**, 36, 1. — [1d] P. De Shong, S. W. Jr. Lander, J. M. Leginus, C. M. Dicken, *Advances in Cycloaddition* (Ed.: D. P. Curran), JAI Press: London **1988**, p. 87. — [1e] R. Annunziata, M. Cinquini, F. Cozzi, L. Raimondi, *Gazz. Chim. Ital.* **1989**, 119, 253. — [1f] E. Breuer, H. G. Aurich, A. Nielsen, *Nitrones, nitronates and nitriloxides* (Eds.: S. Patai, Z. Rappoport), John Wiley & Sons, New York **1989**, pp. 139–312.
- [2] [2a] J. J. Tufariello, J. P. Tette, *J. Org. Chem.* **1975**, 40, 3866–3869. — [2b] M. Joucla, F. Tonnard, D. Gree, J. Hamelin, *J. Chem. Res. (M)* **1978**, 2901–2909. — [2c] J. J. Tufariello, G. E. Lee, P. A. Senaratne, M. Al-Nuri, *Tetrahedron Lett.* **1979**, 45, 4359–4362. — [2d] M. J. Fray, R. H. Jones, E. J. Thomas, *J. Chem. Soc., Perkin Trans. 1* **1985**, 2753–2761. — [2e] Sk. A. Ali, M. I. M. Wazeer, *Tetrahedron* **1988**, 44, 187–193. — [2f] Sk. A. Ali, M. I. M. Wazeer, *J. Chem. Soc., Perkin Trans. 1* **1988**, 597–605. — [2g] Sk. A. Ali, J. H. Khan, M. I. M. Wazeer, H. P. Perzanowski, *Tetrahedron* **1989**, 45, 5979–5986. — [2h] M. Figueredo, J. Font, P. de March, *Chem. Ber.* **1989**, 122, 1701–1704. — [2i] Sk. A. Ali, M. I. M. Wazeer, Mazhar-Ul-Haque, *Tetrahedron* **1990**, 46, 7207–7218. — [2j] Sk. A. Ali, M. I. M. Wazeer, *J. Chem. Soc., Perkin Trans. 2* **1990**, 1035–1039. — [2k] Sk. A. Ali, H. P. Perzanowski, *J. Chem. Res., Synop.* **1992**, 146–147. — [2l] M. Frederickson, *Tetrahedron* **1997**, 53, 403–425. — [2m] K. V. Gothelf, K. A. Jørgensen, *Chem. Rev.* **1998**, 98, 863–909.
- [3] F. M. Cordero, F. Machetti, F. De Sarlo, A. Brandi, *Gazz. Chim. It.* **1997**, 127, 25–29.
- [4] [4a] A. Carriere, A. Virgili, M. Figueredo, *Tetrahedron* **1990**, 46, 2793–2796. — [4b] N. Katagiri, N. Watanabe, J. Sakaki, T. Kawai, C. Kaneko, *Tetrahedron Lett.* **1990**, 31, 4633–4636. — [4c] T. Olsson, K. Stern, G. Westman, S. Sundell, *Tetrahedron* **1990**, 46, 2473–2482. — [4d] S. Murahashi, Y. Imada, M. Kohno, T. Kawakami, *Synlett* **1993**, 395–397. — [4e] B. H. Kim, D. P. Curran, *Tetrahedron* **1993**, 49, 293–318. — [4f] T. Gefflaut, U. Bauer, K. Airola, M. P. Koskinen, *Tetrahedron Asymmetry* **1996**, 7, 3099–3102. — [4g] T. Tejero, A. Dondoni, I. Rojo, A. Merchan, P. Merino, *Tetrahedron* **1997**, 53, 3301–3318.
- [5] For examples of intermolecular cycloaddition to chiral dipolarophiles bearing the stereogenic centre on the ester or amide moiety, see: [5a] Y. Imada, T. Saito, T. Kawakami, S.-I. Murahashi, *Tetrahedron Lett.* **1992**, 33, 5081–5084. — [5b] S. Baskaran, G. K. Trivedi, *J. Chem. Research (S)* **1996**, 12, 542–543. — [5c] K. V. Gothelf, R. G. Hazell, K. A. Jørgensen, *J. Org. Chem.* **1996**, 61, 346–355. — [5d] K. B. Jensen, K. V. Gothelf, R. G. Hazell, K. A. Jørgensen, *J. Org. Chem.* **1997**, 62, 2471–2477.
- [6] For examples of intermolecular cycloaddition to chiral dipolarophiles bearing the stereogenic centre on the acyl moiety, see: [6a] J. Vasu, P. J. Nadkarni, G. K. Trivedi, A. Steigel, *Magn. Reson. Chem.* **1991**, 29, 645–649. — [6b] S. Saito, T. Ishikawa, T. Moriwake, *Synlett* **1994**, 279–281. — [6c] S. Baskaran, G. K. Trivedi, *J. Chem. Research (S)* **1995**, 308–309. — [6d] F. Busqué, P. de March, M. Figueredo, J. Font, M. Monsalvatje, A. Virgili, A. Alvarez-Larena, J. F. Piniella, *J. Org. Chem.* **1996**, 61, 8578–8585. — [6e] S. Baskaran, C. Baskaran, P. J. Nadkarni, G. K. Trivedi, J. Chandrasekhar, *Tetrahedron* **1997**, 53, 7057–7076.
- [7] For examples of cycloaddition to chiral  $\alpha,\beta$ -unsaturated lactones and lactams, see: [7a] M. J. Fray, E. J. Thomas, D. J. Williams, *J. Chem. Soc., Perkin Trans. 1* **1985**, 2763–2767. — [7b] I. Panfil, M. Chmielewski, *Tetrahedron* **1985**, 41, 4713–4716. — [7c] I. Panfil, C. Belzecki, M. Chmielewski, *J. Carbohydrate Chemistry* **1987**, 6, 463–470. — [7d] B. de Lange, B. L. Feringa, *Tetrahedron Lett.* **1988**, 29, 5317–5320. — [7e] I. Panfil, C. Belzecki, M. Chmielewski, K. Suwinska, *Tetrahedron* **1989**, 45, 233–238. — [7f] I. Panfil, C. Belzecki, Z. Uranczyk-Lipkowska, M. Chmielewski, *Tetrahedron* **1991**, 47, 10087–10094. — [7g] A. J. Blake, T. A. Cook, A. C. Forsyth, R. O. Gould, R. M. Paton, *Tetrahedron* **1992**, 48, 8053–8064. — [7h] M. T. Rispens, E. Keller, B. de Lange, W. J. Zijlstra, B. L. Feringa, *Tetrahedron: Asymmetry* **1994**, 5, 607–624. — [7i] N. Langlois, N. Van Bac, N. Dahuron, J.-M. Delcroix, A. Deyne, D. Griffart-Brunet, A. Chiaroni, C. Riche, *Tetrahedron* **1995**, 51, 3571–3586. — [7j] S. Baskaran, G. K. Trivedi, *J. Chem. Research (M)* **1995**, 8, 1853–1864.
- [8] [8a] A. Brandi, F. Cardona, S. Cicchi, F. M. Cordero, A. Goti, in: *Current Trends in Organic Synthesis* (Eds.: C. Scolastico, F. Nicotra), Kluwer Academic/Plenum Publishers, New York, **1999**, pp. 213–220. — [8b] For a review see: J. Cossy, P. Vogel, in: *Studies in Natural Product Chemistry* (Ed.: Atta-ur-Rahman), Elsevier: New York, **1993**; Vol. 12, p. 275.
- [9] A. Schwartz, P. Madan, J. K. Whitesell, R. M. Lawrence, *Org. Synth.* **1990**, 69, 1–9.
- [10] [10a] J. K. Whitesell, C.-L. Liu, C. M. Buchanan, H.-H. Chen, M. A. Minton, *J. Org. Chem.* **1986**, 51, 551–553. — [10b] E. J. Corey, H. E. Hensley, *J. Am. Chem. Soc.* **1975**, 97, 6908–6909.
- [11] [11a] J. K. Whitesell, *Chem. Rev.* **1992**, 92, 953–964. — [11b] G. B. Jones, B. J. Chapman, *Synthesis* **1995**, 475–497.
- [12] J. J. Tufariello, T. J. Tegeler, *Tetrahedron Lett.* **1976**, 45, 4037–4040.
- [13] [13a] F. Bohlmann, *Chem. Ber.* **1958**, 91, 2157–2167. — [13b] T. A. Crabb, R. F. Newton, D. Jackson, *Chem. Rev.* **1971**, 71, 109–126.
- [14] J. F. Maddaluno, N. Gresh, C. Giessner-Prette, *J. Org. Chem.* **1994**, 59, 793–802.
- [15] [15a] B. Mezrhah, F. Dumas, J. d'Angelo, C. Riche, *J. Org. Chem.* **1994**, 59, 500–503. — [15b] F. Dumas, B. Mezrhah, J. d'Angelo, C. Riche, A. Ciaroni, *J. Org. Chem.* **1996**, 61, 2293–2304.
- [16] [16a] W. G. Dauben, R. A. Bunce, *Tetrahedron Lett.* **1982**, 23, 4875–4878. — [16b] A. E. Greene, F. Charbonnier, *Tetrahedron Lett.* **1985**, 26, 5525–5528. — [16c] F. Thiem, R. Rotscheldt, K. E. Breitmaier, *Synthesis* **1989**, 836–843.
- [17] [17a] K. V. Gothelf, I. Thomsen, K. A. Jørgensen, *J. Am. Chem. Soc.* **1996**, 118, 59–64. — [17b] S. Kobayashi, M. Kawamura, *J. Am. Chem. Soc.* **1998**, 120, 5840–5841.
- [18] A. Katsumata, T. Iwaki, K. Fukumoto, M. Ihara, *Heterocycles* **1997**, 46, 605–616.
- [19] The authors have deposited atomic coordinates for this structure with the Cambridge Crystallographic Data Centre. No. CCDC-141378. The coordinates can be obtained free of charge on application to Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1EZ, UK.
- [20] N. Walker, D. Stuart, *Acta Crystallogr. Sect. A* **1983**, 39, 158–166.
- [21] A. Altomare, G. Cascarano, C. Giacovazzo, A. Guagliardi, M. C. Burla, G. Polidori, M. Camalli, *J. Appl. Crystallog.* **1994**, 27, 435.
- [22] G. M. Sheldrick, *SHELXL97: Program for Crystal Structure Refinement*; Institut für Anorganische Chemie der Universität Göttingen. Göttingen, Germany.

Received May 22, 2000  
[O00245]